

New 13-aza Baccatins

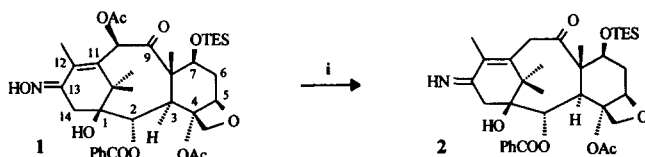
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Abstract: Upon treatment of 7-triethylsilyl-10,13-dideoxy-13-imino baccatin III (**2**) subsequently with diazomethane and *m*-chloroperbenzoic acid a few novel derivatives, namely methylimine **3** and oximes **4** and **5**, were obtained. Interestingly, **4** is characterized by the hydroxyl at position 14.
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In the course of our studies,¹ relative to the modification of the linkage at C-13 between the diterpenoid core and the side chain of paclitaxel, 7-triethylsilyl-10,13-dideoxy-13-imino baccatin III (**2**)² turned out to be a key intermediate for the introduction of a nitrogen atom at C-13. Obtained¹ under Raney-nickel reductive conditions³ from oxime **1** (Scheme 1),⁴ imine **2** showed a remarkable stability that might be explained by the steric hindrance around position 13.⁵

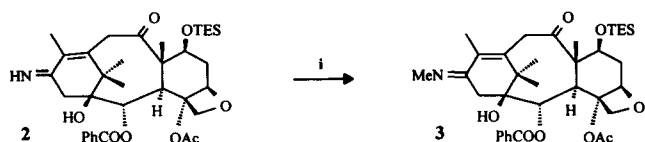
Scheme 1. Synthesis of imine 2



Reagents: i) *W*-2 Raney-nickel, 51% hydrazine hydrate, EtOH, RT, 3h, 57%.

Imine **2** also shows an interesting reactivity which is evident from the examples described below. In particular, when treated with diazomethane in the presence of cuprous bromide, imino baccatin **2** yielded *N*-methyl imine **3**⁶ (Scheme 2) as the major product instead of the somewhat expected aziridine,⁷ a result that has little precedent in the literature.⁸

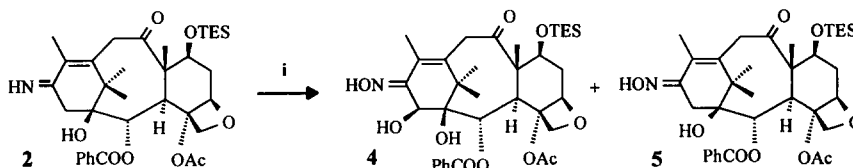
Scheme 2. Synthesis of methylimine 3



Reagents: i) CH₂N₂, diethylether, THF, CuBr, RT, 15', 62%.

When imine **2** was reacted with meta-chloro perbenzoic acid, the two oximes **4** and **5**⁹ were obtained (Scheme 3) as single stereoisomers (configuration not assigned). The formation of the oximes is favoured with respect to the corresponding oxaziridine¹⁰ or nitrone,¹¹ in agreement with the few examples of oxidation of N-H ketimines reported in the literature.^{12,13}

Scheme 3. Synthesis of oximes **4** and **5**



Reagents: *i*) MCPBA, CH₂Cl₂, RT, 40', 40% (4/5 ratio=3:2).

The additional hydroxylation at position 14 in compound **4** resembles the known transformation of enols into α -hydroxy ketones upon treatment with peracids.¹⁴ In fact the tautomeric 13-14 enamine double bond might be epoxidized and subsequently rearranged into the α -hydroxy compound in a similar fashion.¹⁵ Isolation of **4** is particularly interesting because, apart from the 14-hydroxylated taxanes¹⁶ derived from the natural 14 β -hydroxy-10-deacetyl baccatin III,¹⁷ to our knowledge only one 14-OH taxane has been synthesized so far by oxidizing 7-triethylsilyl,10-deacetyl baccatin III with MnO₂.¹⁸

The β configuration of the 14-hydroxyl, established through H-14/H-3, H-14/H-oPh NOESY cross-peaks, is in agreement with the fact that taxanes are cup-shaped and present the concavity toward the α face of the molecule.

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REFERENCES AND NOTES

- Menichincheri, M.; Ceccarelli, W.; Fusar-Bassini, D.; Vanotti, E.; Ciomei, M.; Mongelli, N.; Pinciroli, V.; Vulpetti, A. *Med.Chem.Res.* **1996**, *6*, 264-292.
- 2:** FAB-MS : 640 (M+H)⁺ (C₃₅ H₄₉ NO₈ Si)⁺ (100). ¹H NMR (400MHz, CDCl₃) : δ 8.08 (m, 2H, oPh), 7.60 (m, 1H, pPh), 7.48 (m, 2H, mPh), 5.64 (d, J = 6.6 Hz, 1H, H-2), 4.93 (dd, J = 9.6, 2.2 Hz, 1H, H-5), 4.50 (dd, J = 6.8, 10.6 Hz, 1H, H-7), 4.31 (d, J = 8.2 Hz, 1H, H-20 α), 4.12 (d, J = 8.2 Hz, 1H, H-20 β), 4.07 (d, J = 6.6 Hz, 1H, H-3), 3.93 (d, J = 14.5 Hz, 1H, H-10 α), 3.60 (dq, J = 14.5, 1.2 Hz, 1H, H-10 β), 3.09 (d, J = 18.7 Hz, 1H, H-14 α), 2.68 (d, J = 18.7 Hz, 1H, H-14 β), 2.50 (ddd, J = 14.4, 9.6, 6.8 Hz, 1H, H-6 α), 2.20 (s, 3H, COCH₃), 2.07 (d, J = 1.2 Hz, 3H, H-18), 1.86 (ddd, J = 14.4, 10.6, 2.2 Hz, 1H, H-6 β), 1.59 (s, 3H, H-19), 1.19 (s, 3H, H-16), 1.13 (s, 3H, H-17), 0.95 (t, J = 7.9 Hz, 9H, SiCH₂CH₃), 0.58 (m, 6H, SiCH₂CH₃). ¹³C-NMR (100MHz, CDCl₃): δ 205.5 (C-9), 175.0 (C-13), 169.9 (COCH₃), 166.9 (Ph-COO), 147.6 (C-11), 133.8 (C-pPh) 133.7 (C-12), 130.1 (C-oPh), 129.1 (C-1-Ph), 128.7 (C-mPh), 84.1 (C-5), 80.6 (C-4), 78.8 (C-1), 76.4 (C-20), 73.5 (C-2), 72.0 (C-7), 60.6 (C-8), 46.5 (C-10), 45.7 (C-3), 44.4 (C-15), 39.6 (C-14), 37.3 (C-6), 30.1 (C-17), 22.0 (COCH₃), 20.6 (C-16), 14.8 (C-18), 9.8 (C-19), 6.8 (SiCH₂CH₃), 5.3 (SiCH₂CH₃) [¹H and ¹³C signals were assigned through the analysis of

- NOESY (mixing time=0.7 s), HMQC ($^1J_{\text{HC}}=140$ Hz) and HMBC ($^1J_{\text{HC}}=140$ Hz and $^nJ_{\text{HC}}=7$ Hz) experiments].
- Mignonac, C. *Compt.Rend.* **1920**, *170*, 936; Gurault, Y.; Degouzon, M.; Azzaro, M. *Bull.Soc.Chim.Fr.* **1975**, 386.
 - Obtained as a mixture of (E) and (Z) isomers from the corresponding 13-oxo baccatin.
 - Only some aromatic [Verardo, G.; Giumanini, A.G.; Strazzolini, P.; Poiana, M. *Synth.Comm.* **1988**, *18*, 1501-1511] or sterically hindered ketimines [Guziec, F.S.; Russo, J.M. *Synthesis* **1984**, 479-481] are known to be stable.
 - To a solution of imine **2** (202 mg, 0.316 mmol) in anhydrous THF (60 mL), a 0.4 M solution of diazomethane in ethyl ether (60 mL) was added under nitrogen at room temperature. After 10' CuBr (300 mg, 2.09 mmol) was added and the yellow reaction mixture became dark coloured. After dilution with ethyl ether and filtration, the reaction mixture was concentrated and purified on preparative TLC (n-hexane-ethyl acetate 1:3) to yield 128 mg (62%) of compound **3**.
3: FAB-MS: 654 (M+H)⁺ (C₃₆H₅₁NO₈Si)⁺ (100). ¹H NMR (600 MHz, CDCl₃): δ 8.09 (m, 2H, oPh), 7.62 (m, 1H, pPh), 7.49 (m, 2H, mPh), 5.67 (d, J = 6.3 Hz, 1H, H-2), 4.91 (dd, J = 9.5, 2.0 Hz, 1H, H-5), 4.54 (dd, J = 10.5, 6.8 Hz, 1H, H-7), 4.32 (d, J = 8.5 Hz, 1H, H-20α), 4.17 (d, J = 6.3 Hz, 1H, H-3), 4.12 (d, J = 8.5 Hz, 1H, H-20β), 3.96 (d, J = 14.7 Hz, 1H, H-10α), 3.56 (d, J = 14.7 Hz, 1H, H-10β), 3.31 (s, 3H, NCH₃), 2.75, 2.54 (two d, J = 18.5 Hz, 2H, CH₂-14), 2.49 (ddd, J = 14.3, 9.5, 6.8 Hz, 1H, H-6α), 2.10 (s, 3H, CH₃CO), 207 (s, 3H, H-18), 1.86 (ddd, J = 14.3, 10.5, 2.0 Hz, 1H, H-6β), 1.75 (s, 1H, OH-1), 1.59 (s, 3H, H-19), 1.17 (s, 3H, H-16), 1.05 (s, 3H, H-17), 0.92 (t, J = 8.0 Hz, 9H, SiCH₂CH₃), 0.60 (m, 6H, SiCH₂CH₃). ¹³C NMR (100 MHz, CDCl₃): δ 206.1 (C-9), 169.3 (C=O), 168.2 (C-13), 167.0 (PhCOO), 142.9 (C-11), 135.9 (C-12), 133.8 (C-pPh), 130.0 (C-oPh), 129.1 (C-1-Ph), 128.7 (C-mPh), 84.2 (C-5), 81.0 (C-4), 79.2 (C-1), 76.4 (C-20), 73.7 (C-2), 72.0 (C-7), 60.4 (C-8), 46.2 (C-10), 46.1 (C-3), 43.6 (C-15), 39.4 (NCH₃), 37.3 (C-6), 33.6 (C-14), 29.5 (C-17), 21.7 (CH₃CO-4), 20.6 (C-16), 15.4 (C-18), 9.8 (C-19), 6.8 (SiCH₂CH₃), 5.3 (SiCH₂CH₃). [¹³C signals were assigned through the analysis of gradient enhanced HSQC ($^1J_{\text{HC}}=140$ Hz) and gradient enhanced HMBC ($^1J_{\text{HC}}=140$ Hz and $^nJ_{\text{HC}}=7$ Hz) experiments].
 - Aziridines from N-alkyl iminium salts and diazomethane: Leonard, L.J.; Jann, K.; Paukstelis, J.V.; Steinhardt, C.K. *J.Org.Chem.* **1963**, *28*, 1499-1506; from N-alkylimines and phenyl diazomethane in the presence of CuBr: Bartnik, R.; Mloston, G. *Tetrahedron* **1984**, *40*, 2569-2576; from imines with ethyldiazoacetate (EDA) and chiral Cu(I)-bis(dihydrooxazole) complexes: Hansen, K.B.; Finney, N.S.; Jacobsen, E.N. *Angew.Chem.Int.Ed.Engl.* **1995**, *34*, 676-678; from imines with EDA and Lewis acids: Casarrubios, L.; Pérez, J.A.; Brookhart, M.; Templeton, J.L. *J.Org.Chem.* **1996**, *61*, 8358-8359; from imines with (-)-menthyldiazoacetate and Cu(OTf)₂: Rasmussen, K.G.; Jorgensen, K.A. *J.Chem.Soc., Chem.Comm.* **1995**, 1401-1402.
 - N-alkyl imines from N-H imines using diaryldiazomethane in the presence of bis(acetylacetonato)copper(II): Mehrotra, K.N.; Prasad, G. *Tetrahedron Lett.* **1978**, *43*, 4179-4182.
 - To a solution of imine **2** (55 mg, 0.086 mmol) in CH₂Cl₂ (3 mL) 50% MCPBA was added (52 mg, 0.151 mmol) at room temperature under nitrogen. After stirring for 40' the reaction mixture was diluted with CH₂Cl₂ (10 mL) and extracted twice with sat. NaHCO₃. The organic solution was washed with brine and dried over Na₂SO₄. After filtration and evaporation, the crude material was purified over

preparative TLC (n-hexane-ethyl acetate 1:1) to afford 14 mg (24%) of compound **4** and 9 mg (16%) of compound **5**.

4: FD-MS: 671 (M^{+}) ($C_{35}H_{49}NO_{10}Si$) $^{+}$. 1H NMR (600 MHz, $CDCl_3$): δ 9.0 (bs, 1H, NOH), 8.08 (m, 2H, oPh), 7.59 (m, 1H, pPh), 7.48 (m, 2H, mPh), 5.82 (d, $J = 6.8$ Hz, 1H, H-2), 4.98 (d, $J = 1.8$ Hz, 1H, H-14), 4.90 (dd, $J = 9.6, 2.1$ Hz, 1H, H-5), 4.75 (d, $J = 1.8$ Hz, 1H, OH-14), 4.46 (dd, $J = 10.6, 6.9$ Hz, 1H, H-7), 4.28 (d, $J = 8.5$ Hz, 1H, H-20 α), 4.24 (d, $J = 8.5$ Hz, 1H, H-20 β), 3.96 (s, 1H, OH-1), 3.92 (d, $J = 14.5$ Hz, 1H, H-10 α), 3.89 (d, $J = 6.8$ Hz, 1H, H-3), 3.57 (d, $J = 14.5$ Hz, 1H, H-10 β), 2.47 (ddd, $J = 14.5, 9.6, 6.9$ Hz, 1H, H-6 α), 2.25 (s, 3H, CH_3CO), 2.05 (s, 3H, H-18), 1.88 (ddd, $J = 14.5, 10.6, 2.1$ Hz, 1H, H-6 β), 1.64 (s, 3H, H-19), 1.20 (s, 3H, H-16), 1.11 (s, 3H, H-17), 0.96 (t, $J = 8.0$ Hz, 9H, $SiCH_2CH_3$), 0.58 (m, 6H, $SiCH_2CH_3$). ^{13}C NMR (100 MHz, $CDCl_3$): δ 205.2 (C-9), 170.1 ($COCH_3$), 165.7 (Ph COO), 159.1 (C-13), 146.7 (C-11), 133.3 (C-pPh), 130.0 (C-oPh), 129.7 (C-1-Ph), 128.7 (C-12), 128.5 (C-mPh), 84.1(C-5), 81.1 (C-4), 76.4 (C-20), 75.9 (C-1), 72.0, 71.9 (C-7, C-2), 63.9 (C-14), 60.8 (C-8), 46.0 (C-10), 45.3 (C-3), 43.8 (C-15), 37.3 (C-6), 31.2 (C-17), 21.8 (CH_3CO), 21.3 (C-16), 15.6 (C-18), 9.8 (C-19), 6.8 ($SiCH_2CH_3$), 5.4 ($SiCH_2CH_3$). [1H and ^{13}C signals were assigned through the analysis of NOESY (mixing time = 0.7 s), HMQC ($^1J_{HC} = 140$ Hz) and HMBC ($^1J_{HC} = 140$ Hz and $^nJ_{HC} = 7$ Hz) experiments].

5: FD-MS: 655 (M^{+}) ($C_{35}H_{49}NO_9$) $^{+}$. 1H NMR (400 MHz, $CDCl_3$): δ 8.2 (bs, 1H, NOH), 8.10 (m, 2H, oPh), 7.61 (m, 1H, pPh), 7.49 (m, 2H, mPh), 6.09 (dd, $J = 6.5, 1.0$ Hz, 1H, H-2), 4.94 (dd, $J = 9.5, 2.1$ Hz, 1H, H-5), 4.50 (dd, $J = 10.5, 6.9$ Hz, 1H, H-7), 4.32 (d, $J = 8.3$ Hz, 1H, H-20 α), 4.12 (d, $J = 8.3$ Hz, 1H, H-20 β), 4.01 (d, $J = 6.5$ Hz, 1H, H-3), 3.93 (d, $J = 14.4$ Hz, 1H, H-10 α), 3.54 (dq, $J = 14.4, 1.1$ Hz, 1H, H-10 β), 3.02 (d, $J = 19.7$ Hz, 1H, H-14 α), 2.78 (dd, $J = 19.7, 1.0$ Hz, 1H, H-14 β), 2.49 (ddd, $J = 14.4, 9.5, 6.9$ Hz, 1H, H-6 α), 2.23 (s, 3H, CH_3CO), 2.05 (d, $J = 1.1$ Hz, 3H, H-18), 1.86 (ddd, $J = 14.4, 10.5, 2.1$ Hz, 1H, H-6 β), 1.72 (s, 1H, OH-1), 1.59 (s, 3H, H-19), 1.18 (s, 3H, H-16), 1.09 (s, 3H, H-17), 0.95 (t, $J = 8.0$ Hz, 9H, $SiCH_2CH_3$), 0.59 (m, 6H, $SiCH_2CH_3$). [1H signals were assigned through the analysis of a NOESY experiment (mixing time = 0.7 s)].

10. Patai, S. *The Chemistry of the Carbon-Nitrogen Double Bond*; Wiley : New York, 1970.
11. Ogata, Y.; Sawaki, Y. *J. Am. Chem. Soc.* **1973**, *95*, 4692-4698.
12. With dimethyldioxirane: Boyd, D.R.; Coulter, P.B.; McGuckin, M.R.; Sharma, N.D.; Jennings, W.B.; Wilson, V.E. *J. Chem. Soc., Perkin Trans.* **1990**, *1*, 301-306.
13. With oxygen: Lingelbach, P.; Mueller, U.; Rieber, N.; Witzel, T. *Eur. Pat. Appl.*, **688764** (1995).
14. Horiguchi, Y.; Nakamura, E.; Kuwajima, I. *J. Am. Chem. Soc.* **1989**, *111*, 6257-6265.
15. In the case of the 13-N-acetylimine, obtained from acetylation of **2**, the corresponding 13-14 enaminic tautomer has been isolated.¹
16. Ojima, I.; Park, Y.H.; Sun, C.-M.; Fenoglio, I.; Appendino, G.; Pera, P.; Bernacki, R.J. *J. Med. Chem.* **1994**, *37*, 1408-1410; (b) Kant, J.; Farina, V.; Fairchild, C.; Kadow, J.F.; Langley, D.R.; Long, B.H.; Rose, W.C.; Vyas, D.M. *Bioorganic & Medicinal Chemistry Letters* **1994**, *4*, 1565-1570; (c) Ojima, I.; Fenoglio, I.; Park, Y.H.; Pera, P.; Bernacki, R.J. *Bioorganic & Medicinal Chemistry Letters* **1994**, *4*, 1571-1576.
17. Appendino, G.; Gariboldi, P.; Gabetta, B.; Pace, R.; Bombardelli, E.; Viterbo, D. *J. Chem. Soc., Perkin Trans I* **1992**, 2925-2929.
18. Appendino, G., personal communication.

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